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valence electrons.

Electrons on the
outermost energy
level of an atom.

electron dot

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that show valence electrons as dots. octet rule. Chemical compounds tend to form so that each atom, by gaining, losing, or sharing electrons, has an octet of electrons in its highest occupied ...

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between the
molecules are weak,
so the molecules are
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around. In a solid the
forces

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1) calculate moles of each element (if given mass) 2) divide each mole quantity by the smallest mole value from step 1. 3) use whole number values

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from step 2 as the subscripts to the original equation to write the empirical formula. Percent Composition. -the relative amounts of each element in a compound.

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Electrons on the
outermost energy
level of an atom.

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that show valence electrons as dots. octet rule. Chemical compounds tend to form so that each atom, by gaining, losing, or sharing

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Choice Identify the choice that best completes the statement or answers the question. ____ 1.

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temperature of 200.0
g of aluminum by
10 °C? cal (specific
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