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Nitrides | High Oxides

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representation of the sp^3 hybrid orbital. The energy required to accomplish the sp^3 hybridization and raise the carbon atom from the ground state to the corresponding valence state V , is 230 kJ mol^{-1} .

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provides a complete review of the structure properties, processing and applications of refractory carbides and nitrides. The contents include: the refractory carbides; interstitial carbides, structure and composites; titanium, zirconium, and hafnium carbides; vanadium, niobium and

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are reviewed together in one book because : □ They form two families of closely related materials which have similar atomic structures and chemistry. Carbon and nitrogen are next to each other the second period of the Table of the Elements and, in many cases, the carbides and nitrides

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carbides Nitrides
form solid solutions (known as
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In cemented carbides, typically 60-95
vol.% of the carbide phase is bonded
by a more ductile metal, invariably a

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cobalt, nickel or iron alloy. The most important refractory carbides are WC, TiC and TaC; annual world production of these carbides is about 20 kt, 1.5 kt and 0.35 kt, respectively.

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